

solution definition in chemistry thoughtco - definition of solution in chemistry for solutions with components in the same phase the substances present in lower concentration are solutes while the substance present in highest abundance is the solvent using air as an example oxygen and carbon dioxide gases are solutes while nitrogen gas is the solvent, **solution chemistry chemistry encyclopedia water** - solution chemistry the molecules of sugar as well as those of water must be in constant motion in the solution in the case of liquid solutions the sugar molecules do not move very far before they encounter other molecules diffusion in a liquid is therefore less rapid than diffusion in a gas, **solution chemistry article about solution chemistry** - solution in chemistry homogeneous mixture mixture in chemistry a physical combination of two or more pure substances i e elements or compounds a mixture is distinguished from a compound which is formed by the chemical combination of two or more pure substances in a fixed definite proportion click the link for more information, **solution chemistry britannica com** - solution in chemistry homogenous mixture of two or more substances in relative amounts that can be varied continuously up to what is called the limit of solubility the term solution is commonly applied to the liquid state of matter but solutions of gases and solids are possible, **solution chemistry pennsylvania state university** - solutions when the components of a mixture are uniformly intermingled or when a mixture is homogeneous it is called a solution aqueous solutions those containing water are the most common solutions there are several to define the concentration of a solution depending on the application, **solute definition and examples in chemistry thoughtco** - a solute is defined as the substance that is dissolved in a solution for solutions of fluids the solvent is present in greater amount than the solute concentration is a measurement of the amount of solute present in a chemical solution with respect to the amount of solvent

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